

# Principali attività di ricerca in corso e possibili sviluppi

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*Dipartimento di Ingegneria Meccanica e Aerospaziale*

***Incontro promozione attività di ricerca DIMA***

Roma, 6 febbraio 2017



**SAPIENZA**  
UNIVERSITÀ DI ROMA

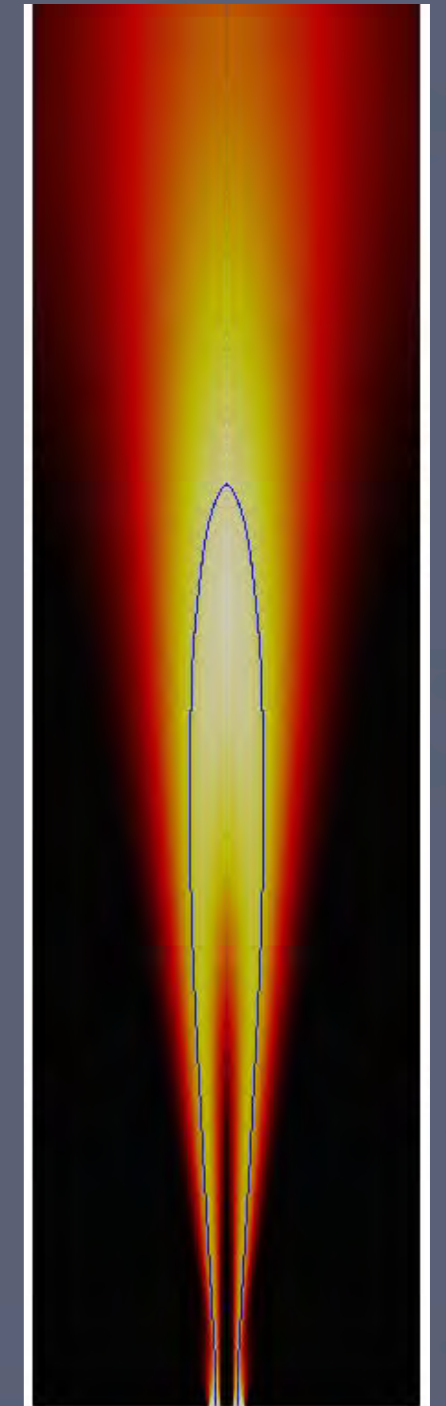
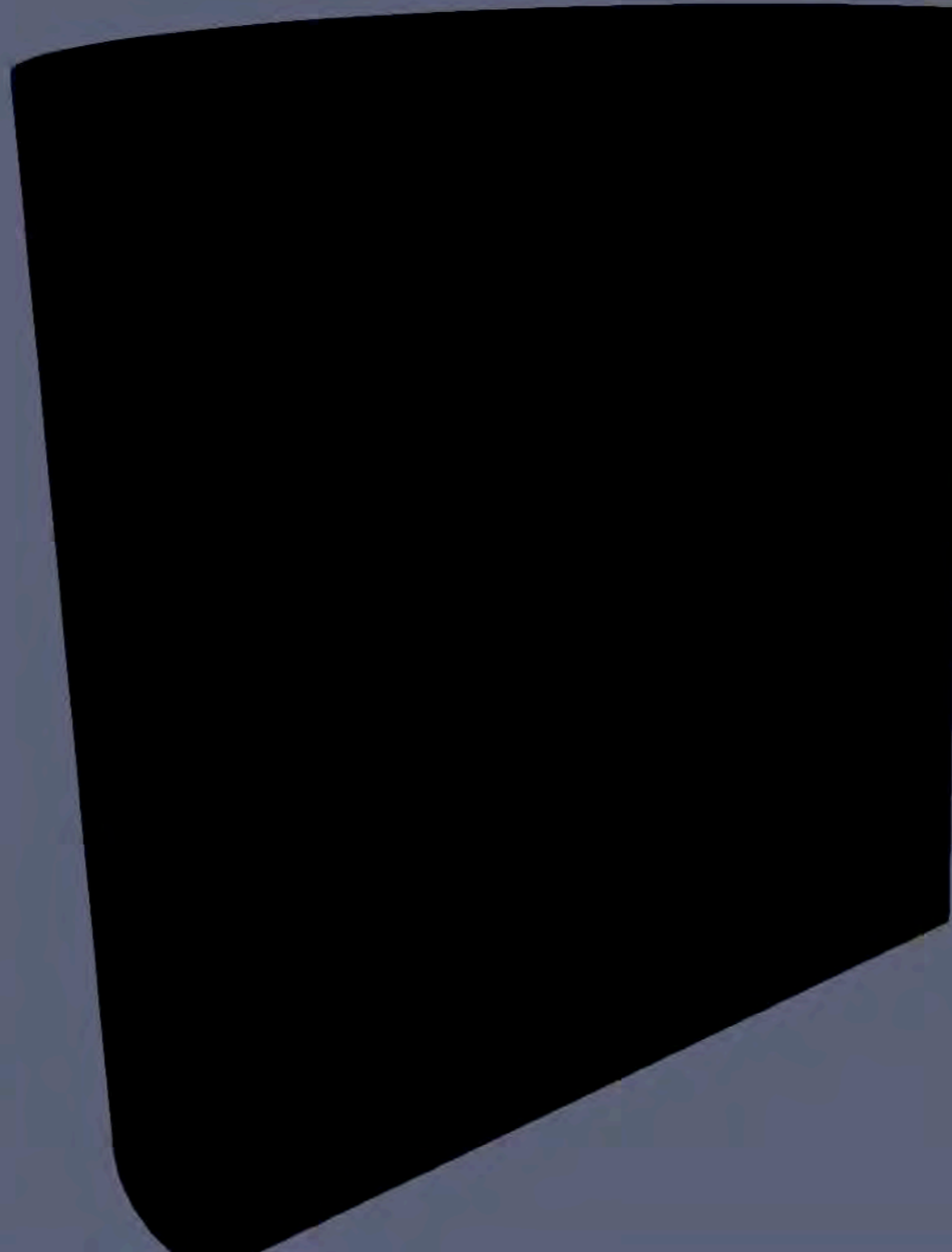
## *MULTI-SCALE SIMULATION OF REACTIVE FLOWS IN THRUST CHAMBERS*

- TURBULENT COMBUSTION CLOSURE (LES VIA CONDITIONAL MOMENT CLOSURE)
- IGNITION TRANSIENT IN COMBUSTION CHAMBERS (URANS)
- SUPERCRITICAL COMBUSTION IN LRE CHAMBERS (RANS, URANS, LES)
- DETERMINISTIC AND PROBABILISTIC MODEL REDUCTION OF CHEMICAL KINETICS USING CSPTK
- PREDICTION OF STRONG AND WEAK IGNITION REGIMES IN TURBULENT REACTING FLOWS WITH TEMPERATURE FLUCTUATIONS
- FLAME INITIATION AND FORMATION (DNS VIA WAVELETS)
- SPRAY MODELING OF PRESSURE-SWIRL ATOMIZERS USING OPENFOAM
- DEVELOPMENT OF THE CSP TOOL KIT (CSPTK) LIBRARY

### *AN EARLY CASE OF CRO-SSD COLLABORATION (06+07)*

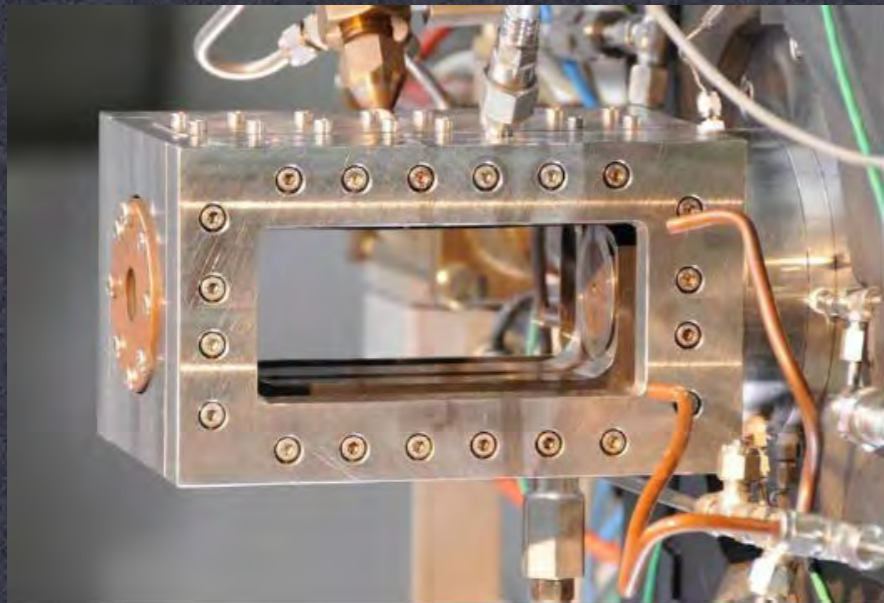
- DETACHED EDDY SIMULATION SHOCK UNSTEADINESS IN AN OVER-EXPANDED PLANAR NOZZLE

MAIN TOPICS



Target test case: Sandia flame A jet diffusion flames ([www.sandia.gov/TNF/DataArch/FlameA/SandiaPilotDoc21.pdf](http://www.sandia.gov/TNF/DataArch/FlameA/SandiaPilotDoc21.pdf))

**TURBULENT COMBUSTION MODELING OF LABSCALE FLAMES (KAUST)**  
OPENFOAM LES (P.P.CIOTTOLI, P.E.LAPENNA, F.CRETA, M.VALORANI)



The experiment consists of triggering ignition in a well controlled gas/gas injection configuration, using a laser beam in order to control the ignition point location and energy release

The flame kernel propagates downstream near the symmetry axis as convected by the fast central jet

The flame kernel propagates across the recirculation region away from the axis

When methane is not entering the chamber anymore, the cold oxygen jet is not consumed and leaves the chamber unburned-> the temperature field at the axis becomes very cold

A significant amount of hot products is still present in the chamber ready to reignite the propellants when the chamber pressure is lowered by the mass loss through the choked nozzle

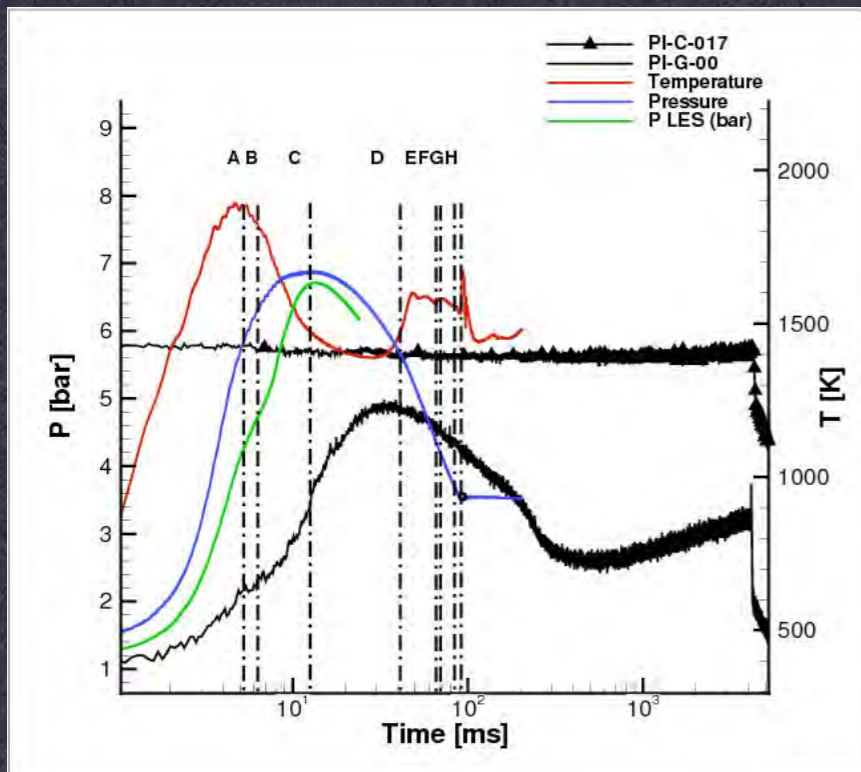
## Pressure



## Temperature

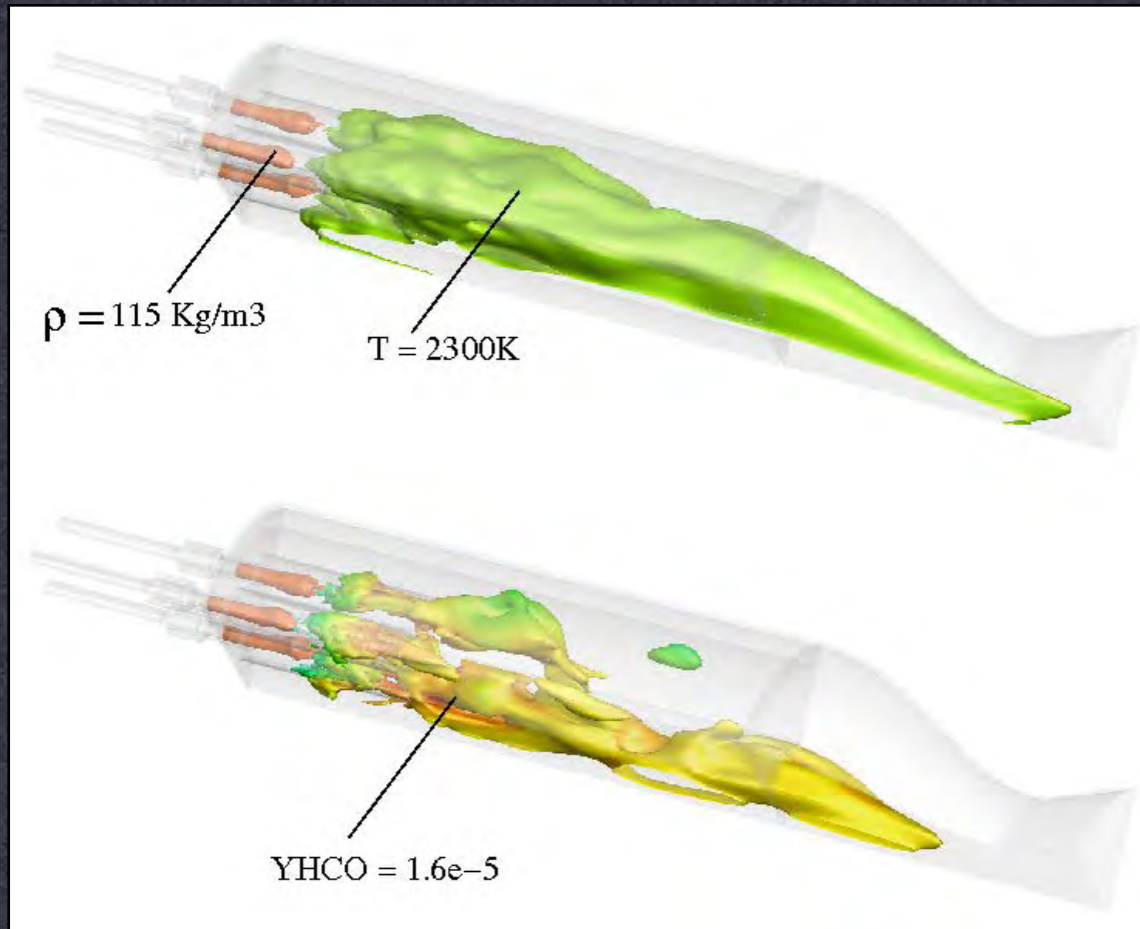


## HCO



# IGNITION TRANSIENT IN COMBUSTION CHAMBERS

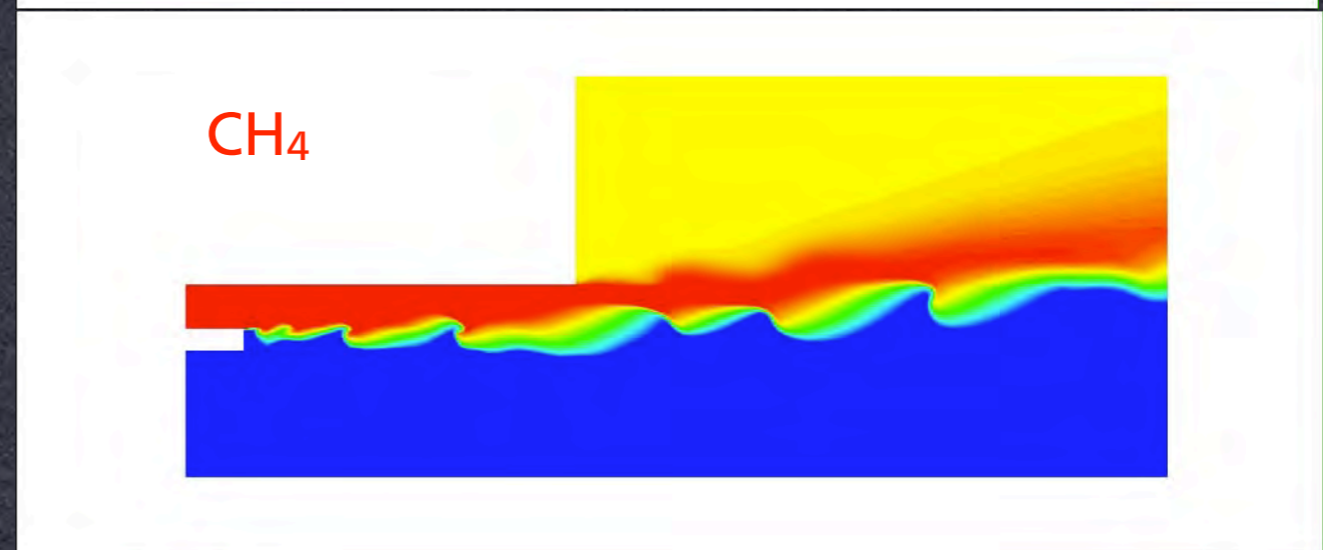
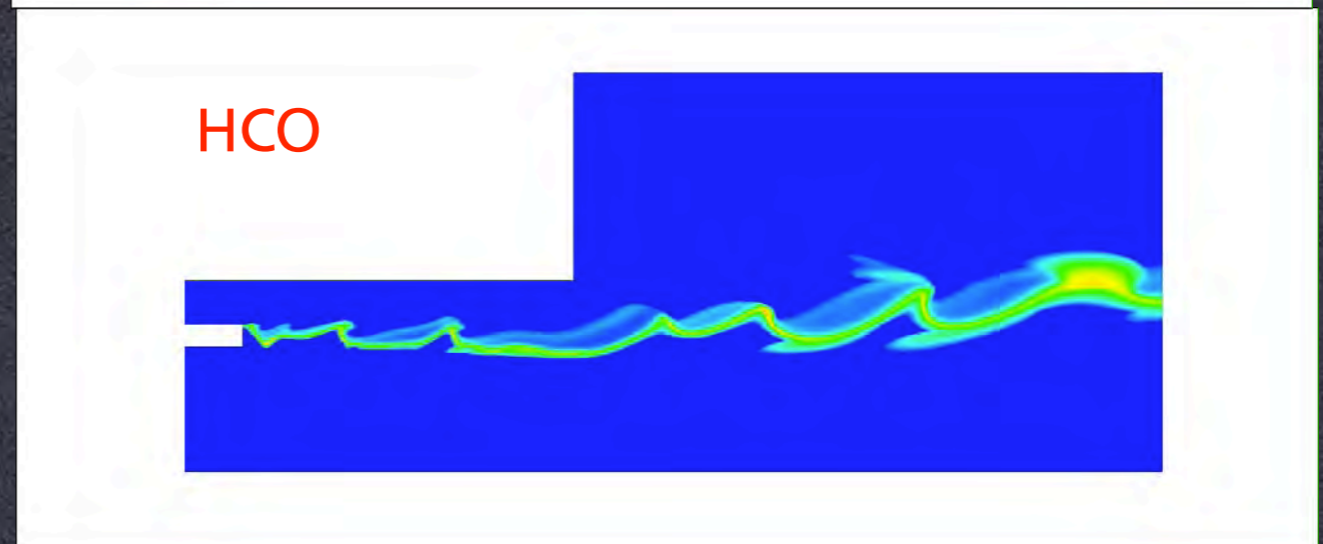
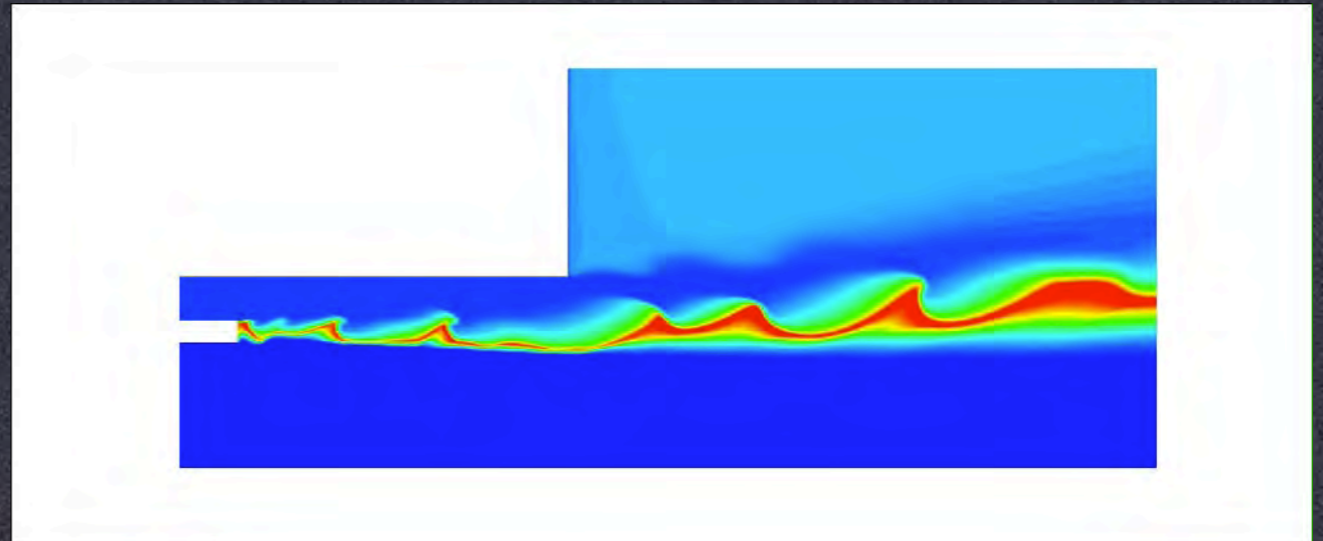
URANS (M.VALORANI, E.MARTELLI, P.P.CIOTTOLI, G.GARGIULO)



Supercritical combustion involves real gas equations of state.

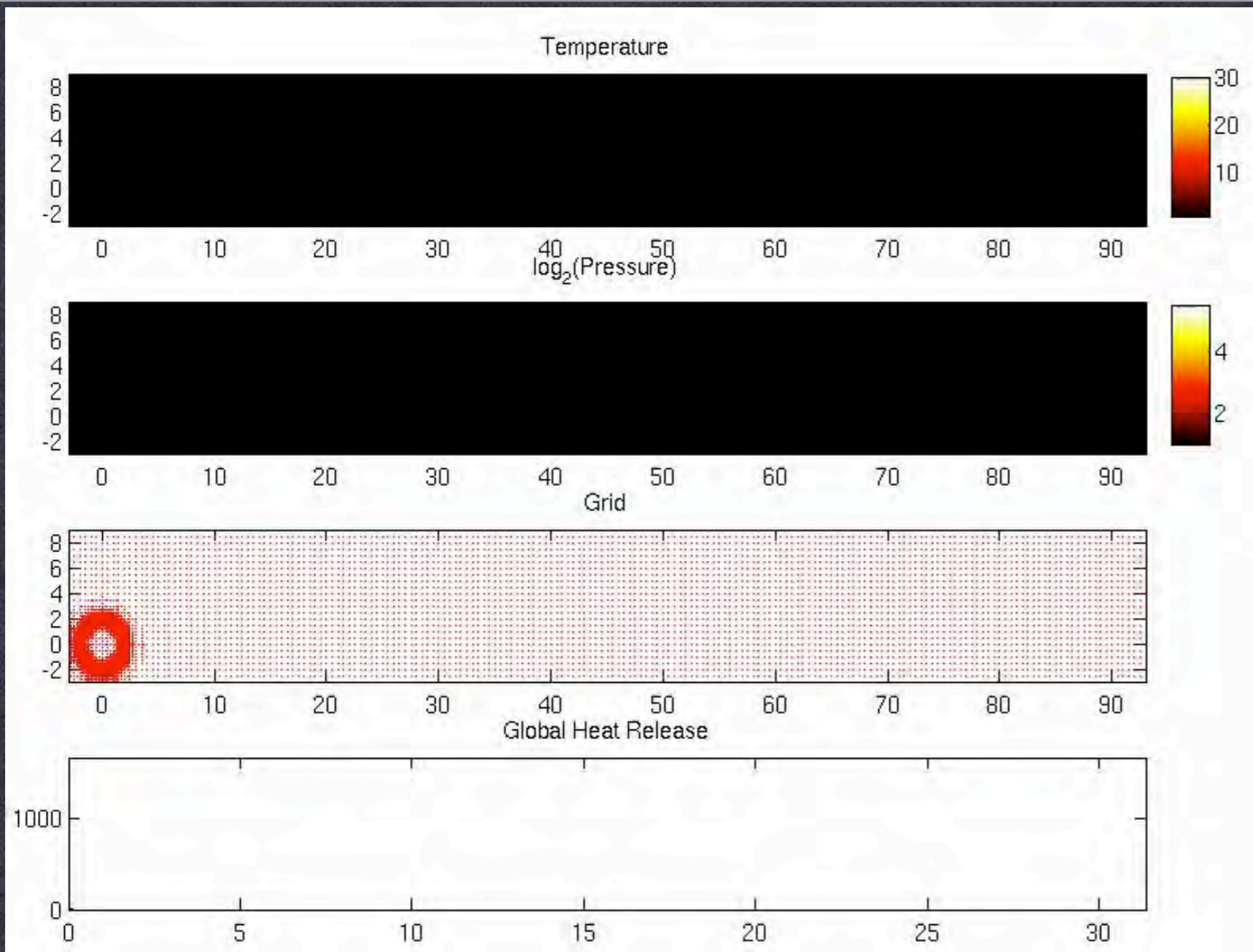
Very strong density gradients occur in the mixing region formed by cryogenic oxygen and gaseous methane.

Anchoring is hindered by the high heat transfer from the flame to the cold liquid oxygen jet



## SUPERCritical COMBUSTION IN LRE CHAMBERS

RANS, URANS, LES (G.GARGIULO, F.CRETA, M.VALORANI)



IGNITION IN A CAVITY

Flames are characterized by a large spectrum of spatial and temporal scales  $\Rightarrow$  multi-scale problems

Direct Numerical Simulations (DNS) involves a very large number of unknowns (if a uniform mesh is used)

Dynamically adaptive wavelet collocation algorithms are ideally suited for DNS of combustion with realistic chemistry:

- the state functions are represented in terms of multiscale basis functions (scaling functions & wavelets), characterized by excellent localization properties both in physical and spectral space.
- adaptation is obtained by retaining only wavelets with an amplitude greater than a user defined threshold, thus providing a verified solution of prescribed accuracy and saving in computer time and memory storage.

## FLAME INITIATION AND FORMATION

“WAVELET ADAPTIVE MESH RESOLUTION” (S.GEMINI, S.PAOLUCCI, E.MARTELLI, M.VALORANI)



# Detached Eddy Simulation shock unsteadiness in an over-expanded planar nozzle

E.Martelli, P.P. Ciottoli, M. Bernardini, F. Nasuti, S. Pirozzoli, M. Valorani

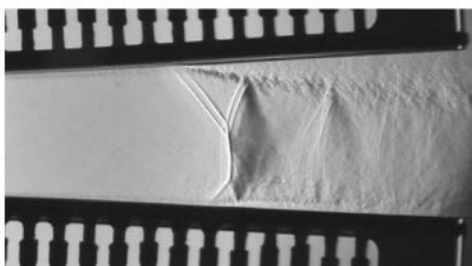


a)  $M_1 = 1.26$

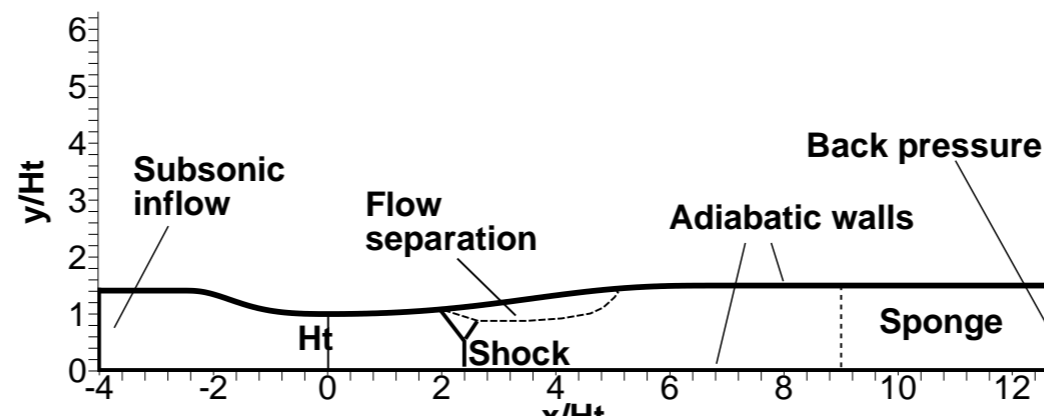


b)  $M_1 = 1.48$

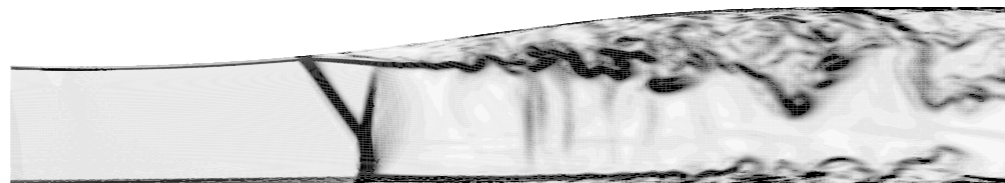
Handa, T., Masuda, M., and Matsuo, K.,



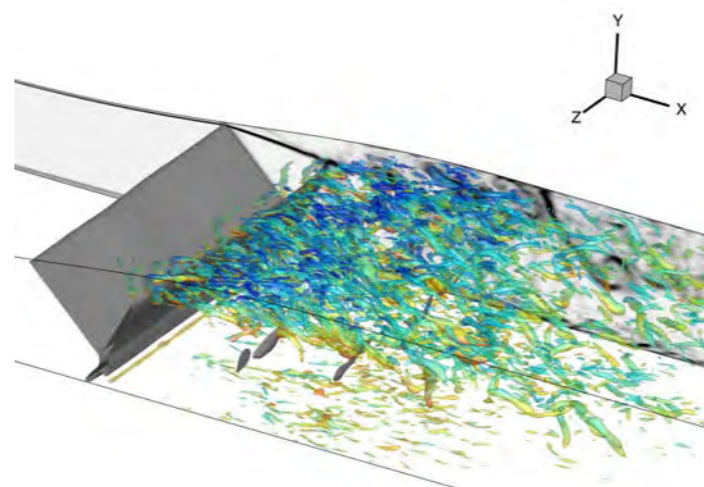
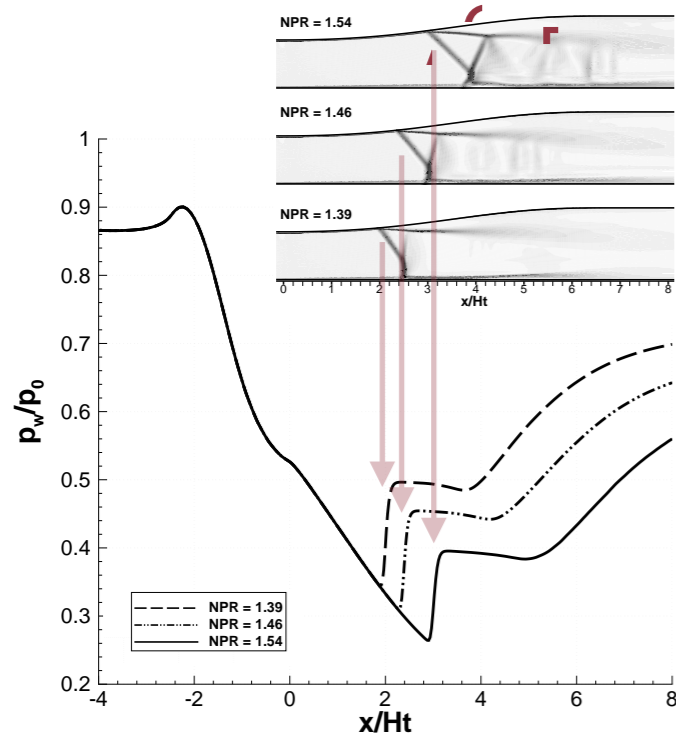
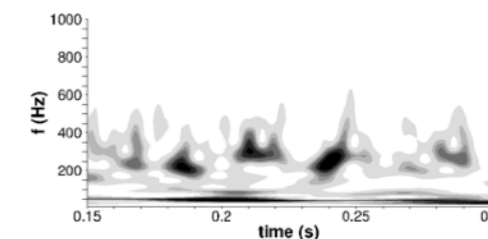
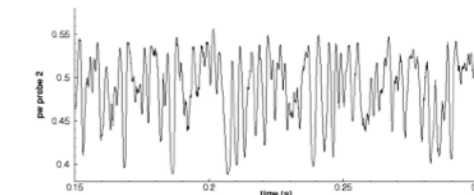
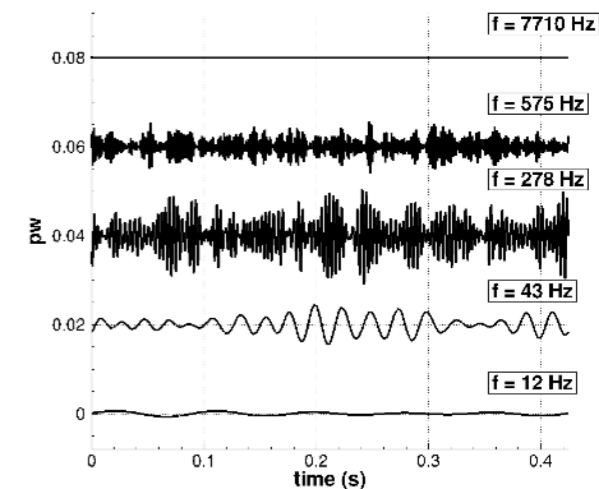
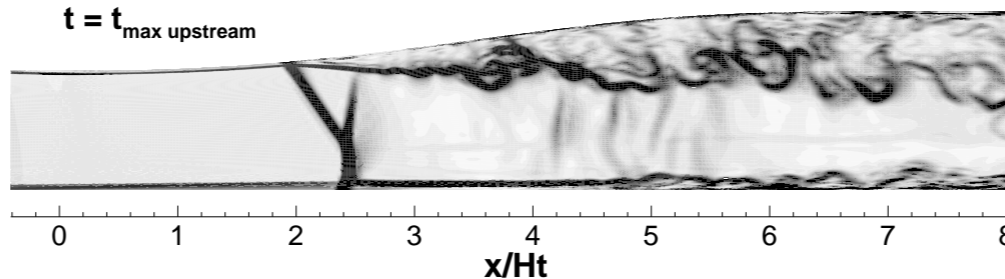
Johnson, A. D. and Papamoschou, D., "Instability



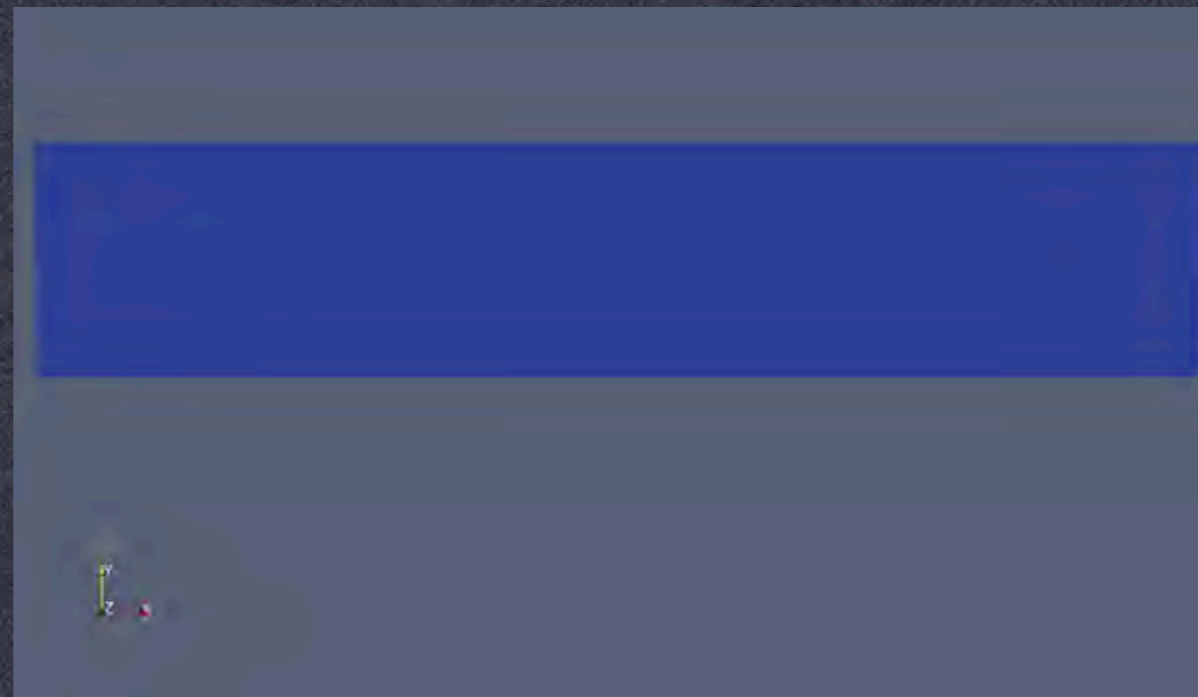
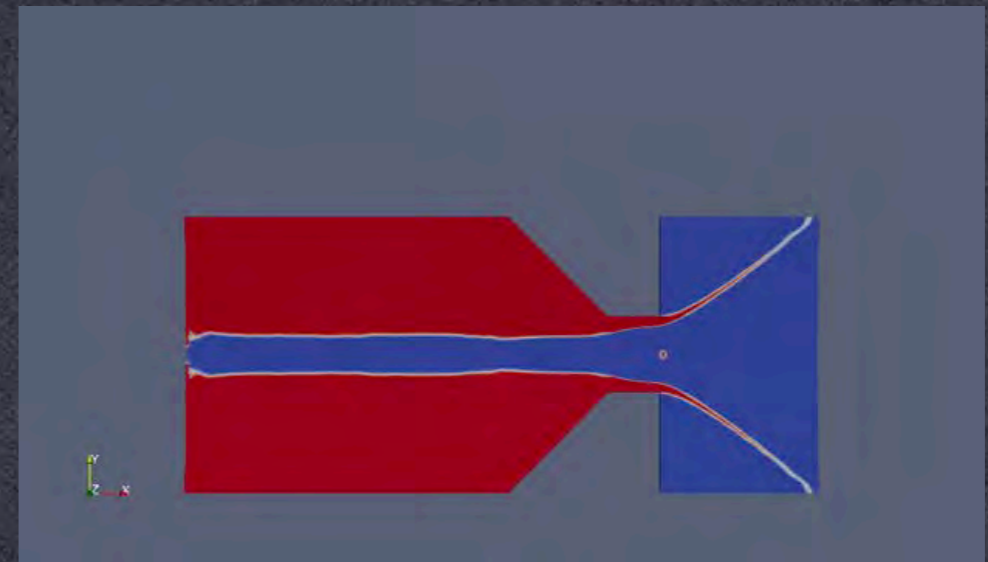
$t = t_{\max \text{ downstream}}$



$t = t_{\max \text{ upstream}}$



The shock movement and the recirculating region have been recognized to be characterized, in the time-frequency space, by a collection of events with a modulation of the oscillation amplitude and a modulation of the frequency



Multi-phase CFD simulations with InterFOAM solver based on the volume of fluid (VoF) method.

Top: Two dimensional axisymmetric simulation of a swirl atomizer

Bottom: Liquid jet breakup detail of the injector

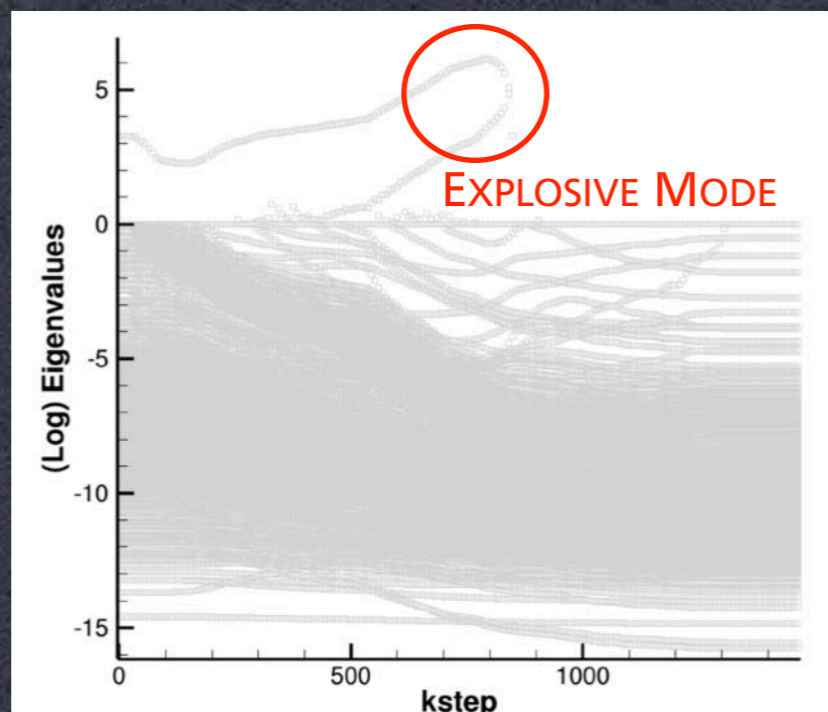
**MULTI-PHASE FLOW: SWIRL ATOMIZER**

OPENFOAM RANS, LES WITH VOF (A. SALEM, P.P.CIOTTOLI, M.VALORANI)

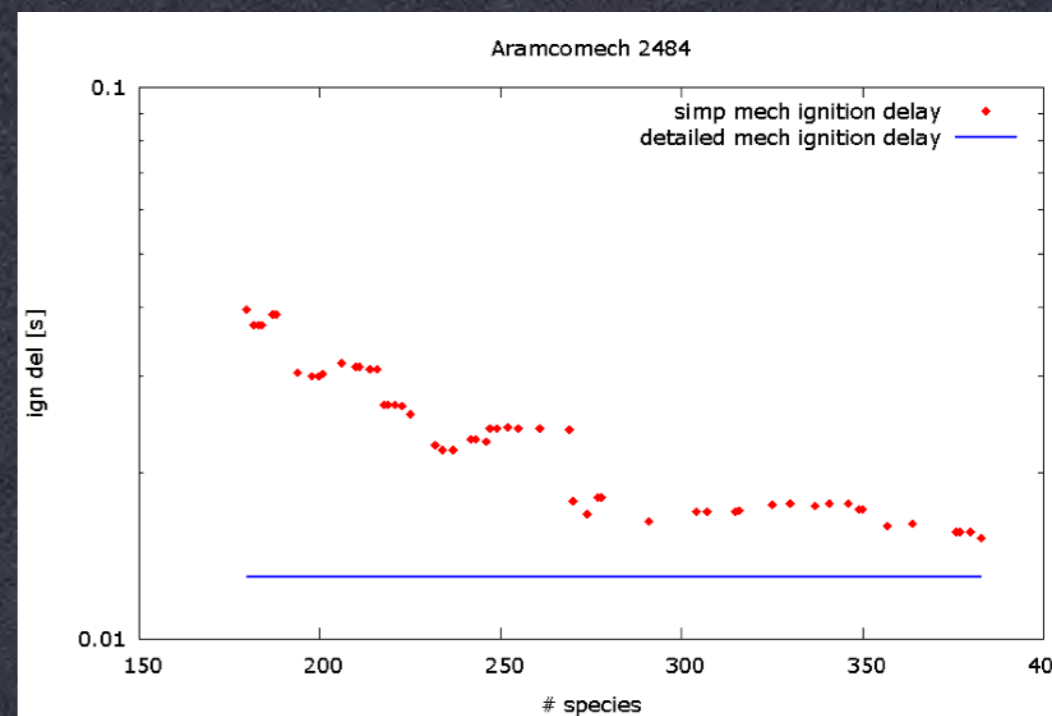


ARAMCO MECH  
2484 SPECIES  
10368 REACTIONS

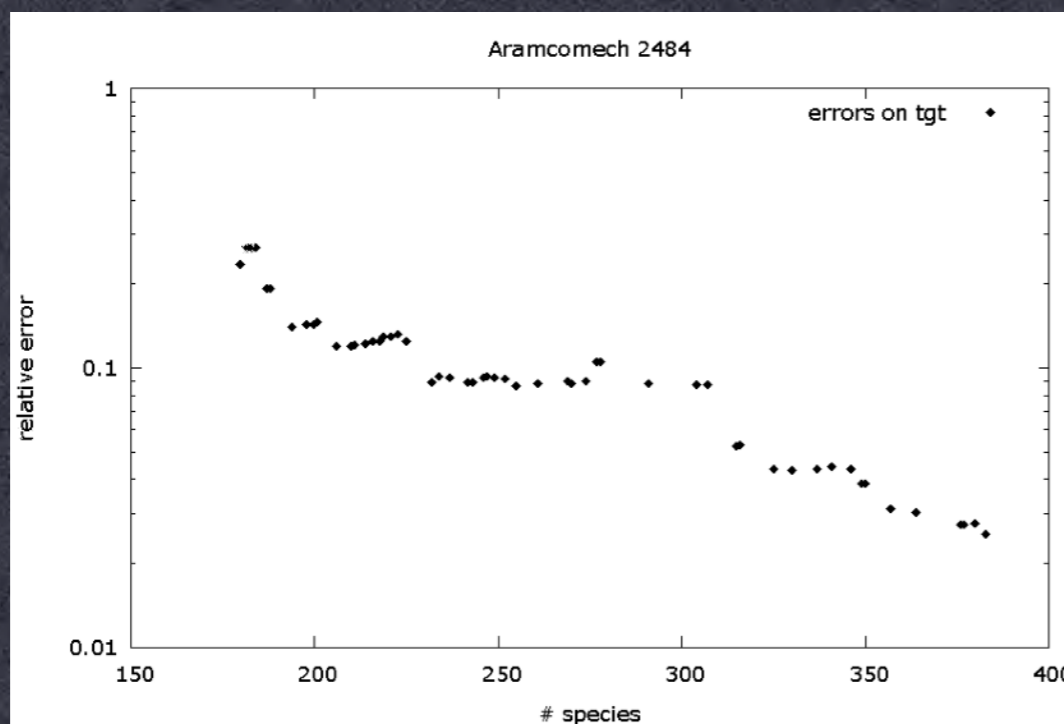
C6H5CH3	0.2923
NC7H16	0.0802
NC4H9OH	0.2756
IC8H18	0.3519



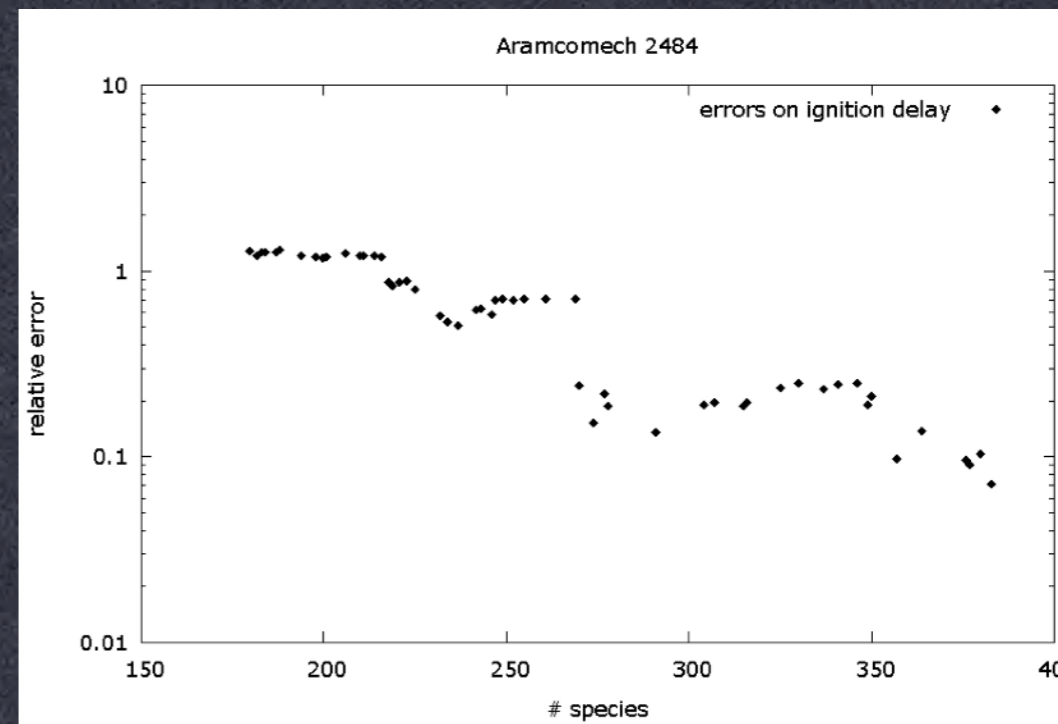
TIME SCALES EVOLUTION



IGNITION DELAY TIME VS DEGREE OF SIMPLIFICATION



ERRORS ON TARGET VS # OF SPECIES



ERRORS ON IGN.TIME VS # OF SPECIES

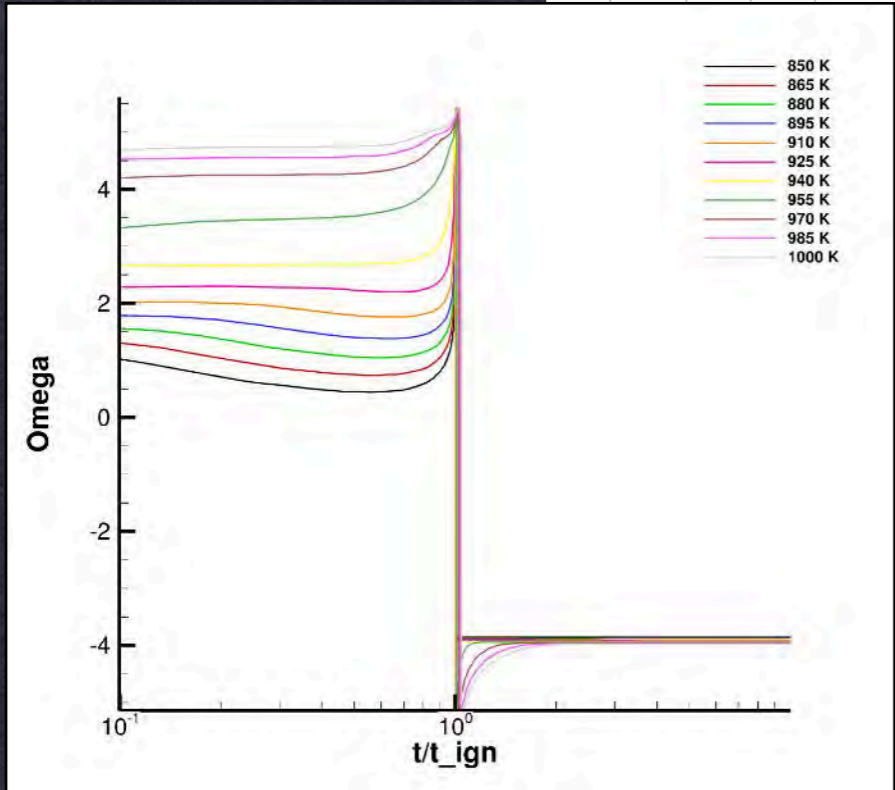
# AUTOMATIC GENERATION OF SIMPLIFIED KINETIC MECHANISMS

M. VALORANI, F. CRETA, H.N. NAJM, D.A. GOUSSIS, P.P. CIOTTOLI, R.MALPICA

**SYNGAS MECH (H<sub>2</sub>+CO/AIR)**  
**57 SPECIES**  
**269 REACTIONS**

**ANALYSIS OF KINETICS ABOUT  
 CROSSOVER TEMPERATURE  
 (950K)**

region #	time @ beginning	T @ beginning	Species	Reactions	Average PI [%]	Bar Chart	region #	time @ beginning	T @ beginning	Species	Reactions	Average PI [%]	Bar Chart
1	0.000513803	900.	H O HO <sub>2</sub>	Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 1 - H+O <sub>2</sub> =>OH+O Rf 28 - CO+O <sub>2</sub> =>CO <sub>2</sub> +O	35.4 30.83 19.17		1	0.	1200.	H OH O	Rb 12 - HO <sub>2</sub> +H<=>H <sub>2</sub> +O <sub>2</sub> Rf 28 - CO+O <sub>2</sub> =>CO <sub>2</sub> +O	58.36 41.38	
2	0.000123912	900.	HO <sub>2</sub>	Rf 1 - H+O <sub>2</sub> =>OH+O Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 28 - CO+O <sub>2</sub> =>CO <sub>2</sub> +O	34.03 33.09 16.		2	8.80654 x 10 <sup>-8</sup>	1200.	H OH O	Rb 12 - HO <sub>2</sub> +H<=>H <sub>2</sub> +O <sub>2</sub> Rf 28 - CO+O <sub>2</sub> =>CO <sub>2</sub> +O Rf 2 - H <sub>2</sub> +O<=>OH+H	50.77 42.23 4.057	
3	0.00262922	900.	HO <sub>2</sub>	Rf 1 - H+O <sub>2</sub> =>OH+O Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M)	36.73 35.69		3	2.74948 x 10 <sup>-7</sup>	1200.	H OH O	Rb 12 - HO <sub>2</sub> +H<=>H <sub>2</sub> +O <sub>2</sub> Rf 28 - CO+O <sub>2</sub> =>CO <sub>2</sub> +O Rf 2 - H <sub>2</sub> +O<=>OH+H	45.27 42.58 5.173 0.6622	
4	0.00901972	900.002	HO <sub>2</sub>	Rf 1 - H+O <sub>2</sub> =>OH+O Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 27 - CO+HO <sub>2</sub> =>CO <sub>2</sub> +OH	38.01 37.03 12.13		4	6.64043 x 10 <sup>-7</sup>	1200.	H O	Rf 1 - H+O <sub>2</sub> =>OH+O Rb 12 - HO <sub>2</sub> +H<=>H <sub>2</sub> +O <sub>2</sub> Rf 28 - CO+O <sub>2</sub> =>CO <sub>2</sub> +O Rf 2 - H <sub>2</sub> +O<=>OH+H	27.79 26.7 26.5 12.18	
5	0.0242431	900.045	HO <sub>2</sub> HO <sub>2</sub>	Rf 1 - H+O <sub>2</sub> =>OH+O Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 27 - CO+HO <sub>2</sub> =>CO <sub>2</sub> +OH	36.32 36.11 11.26		5	4.45471 x 10 <sup>-6</sup>	1200.	H O	Rf 1 - H+O <sub>2</sub> =>OH+O Rf 2 - H <sub>2</sub> +O<=>OH+H	62.88 19.39	
6	0.0345474	900.274	HO <sub>2</sub> HO <sub>2</sub>	Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 1 - H+O <sub>2</sub> =>OH+O	35.96 34.43		6	0.0000441018	1220.55	NaS H O	Rf 1 - H+O <sub>2</sub> =>OH+O Rf 2 - H <sub>2</sub> +O<=>OH+H Rf 3 - H <sub>2</sub> +OH<=>H <sub>2</sub> O+H	51.02 21.17 11.67	
7	0.0435632	900.919	NaS HO <sub>2</sub> HO <sub>2</sub>	Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 1 - H+O <sub>2</sub> =>OH+O	36.42 31.92		7	0.0000472151	1260.31	NaS H O H <sub>2</sub>	Rf 1 - H+O <sub>2</sub> =>OH+O Rf 2 - H <sub>2</sub> +O<=>OH+H Rf 3 - H <sub>2</sub> +OH<=>H <sub>2</sub> O+H	42.53 20.96 13.17	
8	0.0576672	904.771	NaS H HO <sub>2</sub> HO <sub>2</sub>	Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 1 - H+O <sub>2</sub> =>OH+O Rf 11 - H <sub>2</sub> +H<=>H <sub>2</sub> O	37.93 29.58 11.23		8	0.0000477381	1271.88	NaS H O H <sub>2</sub>	Rf 1 - H+O <sub>2</sub> =>OH+O Rf 2 - H <sub>2</sub> +O<=>OH+H Rf 3 - H <sub>2</sub> +OH<=>H <sub>2</sub> O+H Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M)	40.88 20.86 13.3 5.435	
9	0.0584542	905.757	NaS H O HO <sub>2</sub> HO <sub>2</sub>	Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 1 - H+O <sub>2</sub> =>OH+O Rf 11 - H <sub>2</sub> +H<=>H <sub>2</sub> O	38.1 28.71 13.36		9	0.0000479815	1277.92	NaS H O <sub>2</sub> O H <sub>2</sub>	Rf 1 - H+O <sub>2</sub> =>OH+O Rf 2 - H <sub>2</sub> +O<=>OH+H Rf 3 - H <sub>2</sub> +OH<=>H <sub>2</sub> O+H Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M)	38.43 20.6 13.29 6.296	
10	0.0586682	906.306	H O HO <sub>2</sub>	Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 1 - H+O <sub>2</sub> =>OH+O Rf 11 - H <sub>2</sub> +H<=>H <sub>2</sub> O	36.17 27.42 16.2		10	0.0000491772	1314.53	NaS H O <sub>2</sub> O H <sub>2</sub> H <sub>2</sub> O	Rf 1 - H+O <sub>2</sub> =>OH+O Rf 2 - H <sub>2</sub> +O<=>OH+H Rf 3 - H <sub>2</sub> +OH<=>H <sub>2</sub> O+H Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M)	33.3 19.67 12.75 7.762	
11	0.0589165	908.509	H O	Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 1 - H+O <sub>2</sub> =>OH+O Rf 11 - H <sub>2</sub> +H<=>H <sub>2</sub> O	29.9 27.4 18.08		11	0.0000493103	1319.36	NaS H O <sub>2</sub> OH O H <sub>2</sub> H <sub>2</sub> O	Rf 1 - H+O <sub>2</sub> =>OH+O Rb 1 - H+O <sub>2</sub> =>OH+O Rf 2 - H <sub>2</sub> +O<=>OH+H Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 3 - H <sub>2</sub> +OH<=>H <sub>2</sub> O+H	30.39 18.81 12.13 10.23 8.35	
12	0.0590903	915.732	NaS H O	Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 1 - H+O <sub>2</sub> =>OH+O Rf 11 - H <sub>2</sub> +H<=>H <sub>2</sub> O	30.36 24.88 16.6		12	0.0000502275	1356.97	NaS H O <sub>2</sub> OH O H <sub>2</sub> H <sub>2</sub> O CO <sub>2</sub>	Rf 1 - H+O <sub>2</sub> =>OH+O Rb 1 - H+O <sub>2</sub> =>OH+O Rf 2 - H <sub>2</sub> +O<=>OH+H Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 3 - H <sub>2</sub> +OH<=>H <sub>2</sub> O+H	27.05 17.52 11.48 10.5 8.769	
13	0.059041	926.094	NaS H O O	Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 1 - H+O <sub>2</sub> =>OH+O Rf 11 - H <sub>2</sub> +H<=>H <sub>2</sub> O	37.82 13.37 13.27 12.39		13	0.0000503559	1362.81	NaS H O <sub>2</sub> OH O H <sub>2</sub> H <sub>2</sub> O CO CO <sub>2</sub>	Rf 1 - H+O <sub>2</sub> =>OH+O Rb 1 - H+O <sub>2</sub> =>OH+O Rf 2 - H <sub>2</sub> +O<=>OH+H Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 3 - H <sub>2</sub> +OH<=>H <sub>2</sub> O+H	22.94 15.26 12.63 10.64 10.45 8.447	
25	0.0591116	1344.16	NaS H O <sub>2</sub> OH O H <sub>2</sub> H <sub>2</sub> O CO <sub>2</sub>	Rf 1 - H+O <sub>2</sub> =>OH+O Rb 1 - H+O <sub>2</sub> =>OH+O Rf 10 - H+O <sub>2</sub> (M)=>HO <sub>2</sub> (M) Rf 2 - H <sub>2</sub> +O<=>OH+H Rf 26 - CO+OH<=>CO <sub>2</sub> +H Rf 3 - H <sub>2</sub> +OH<=>H <sub>2</sub> O+H	20.81 14.08 13.07 12.68 10.32 9.05		25						

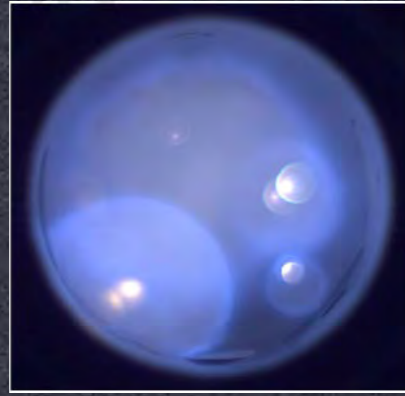


**AUTOMATIC DIAGNOSTICS OF KINETIC MECHANISMS**

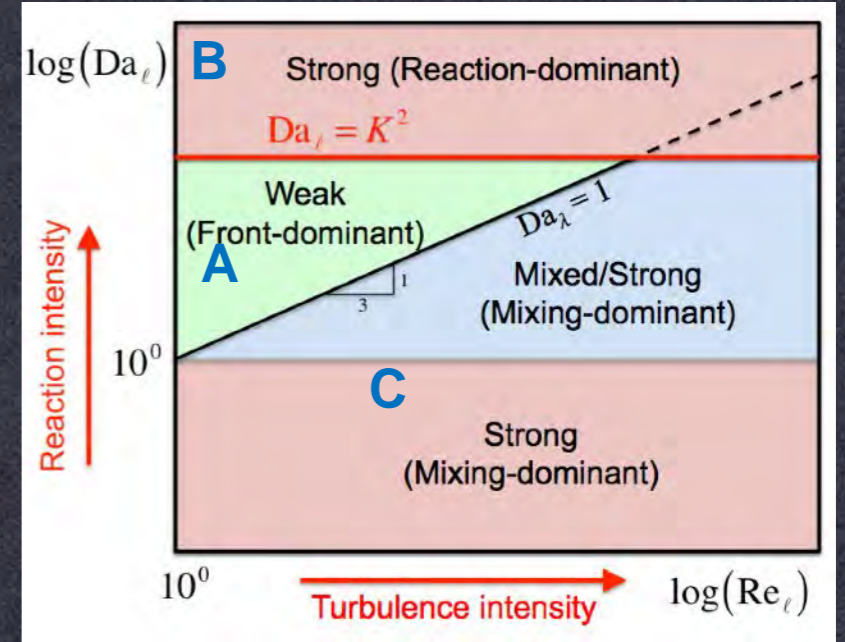
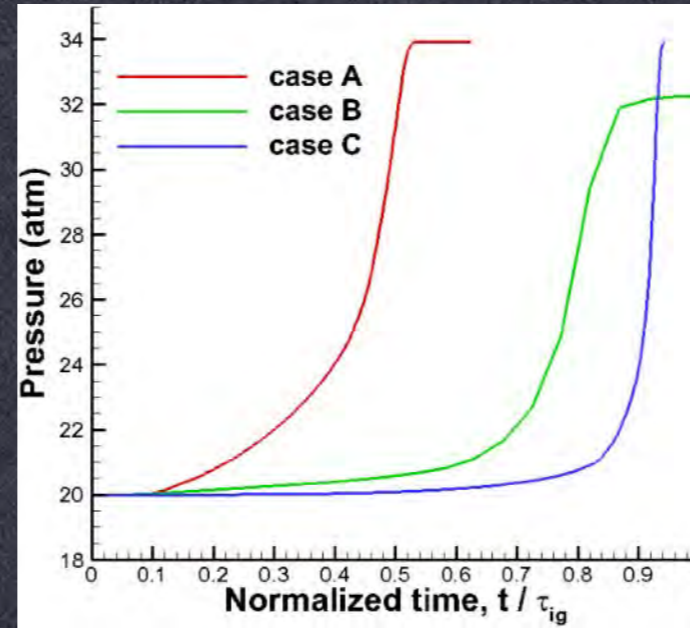
M. VALORANI, P.P.CIOTTOLI, R.MALPICA



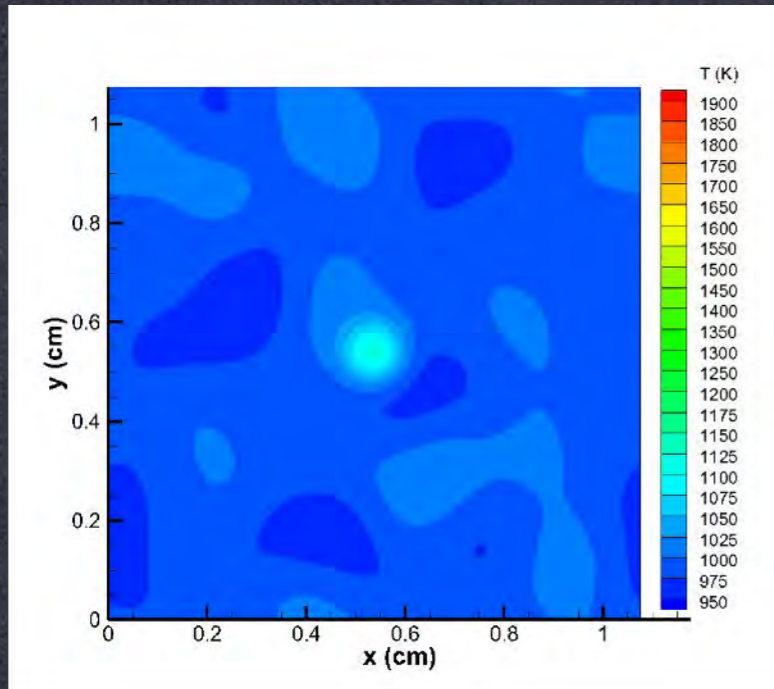
Volumetric ignition (strong)



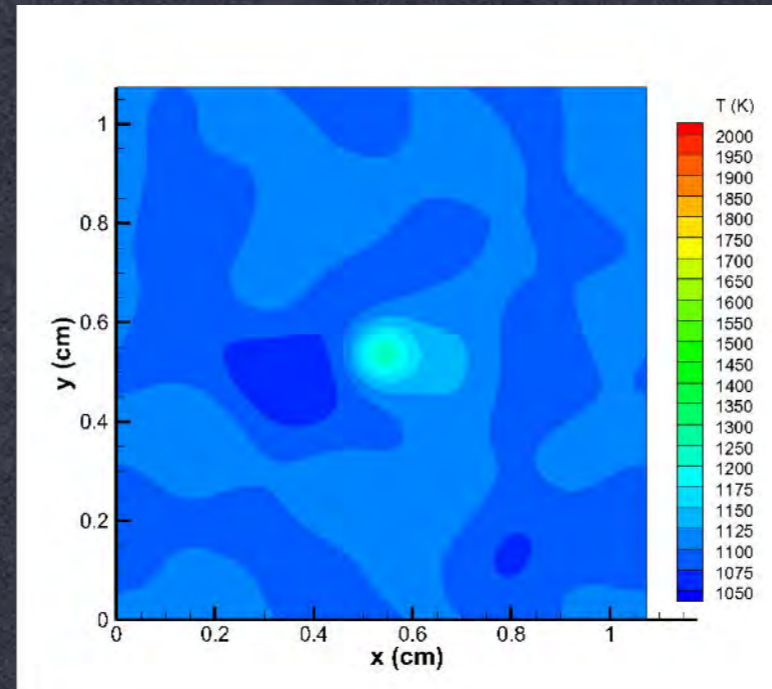
Localized ignition sites + deflagrative fronts (weak)



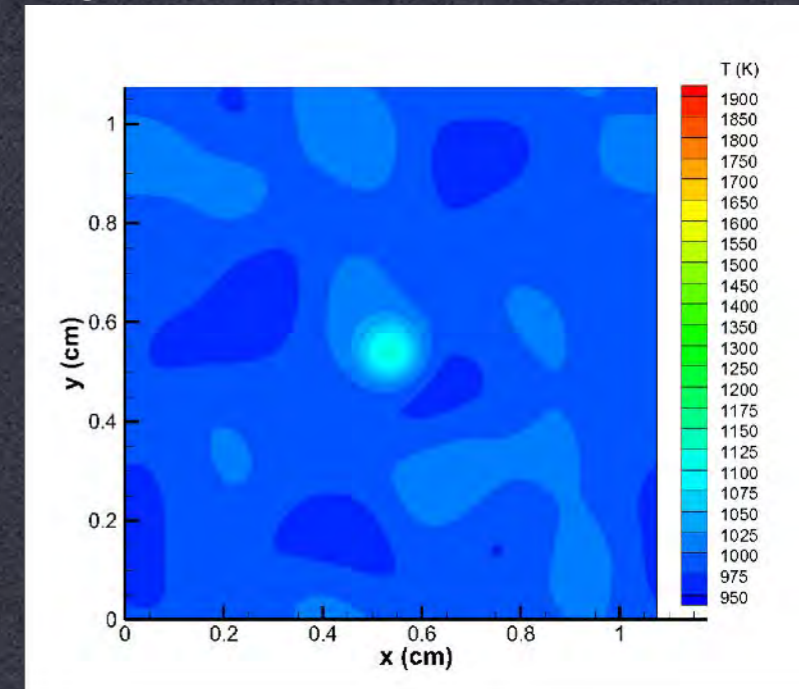
A. Weak Ignition



B. Strong Ignition (Reaction-dominant)



C. Strong Ignition (Mixing-dominant)



PREDICTION OF STRONG AND WEAK IGNITION REGIMES IN TURBULENT REACTING FLOWS WITH TEMPERATURE FLUCTUATIONS

P. PINAKI, H. G. IM, P. P. CIOTTOLI, R. MALPICA, M. VALORANI



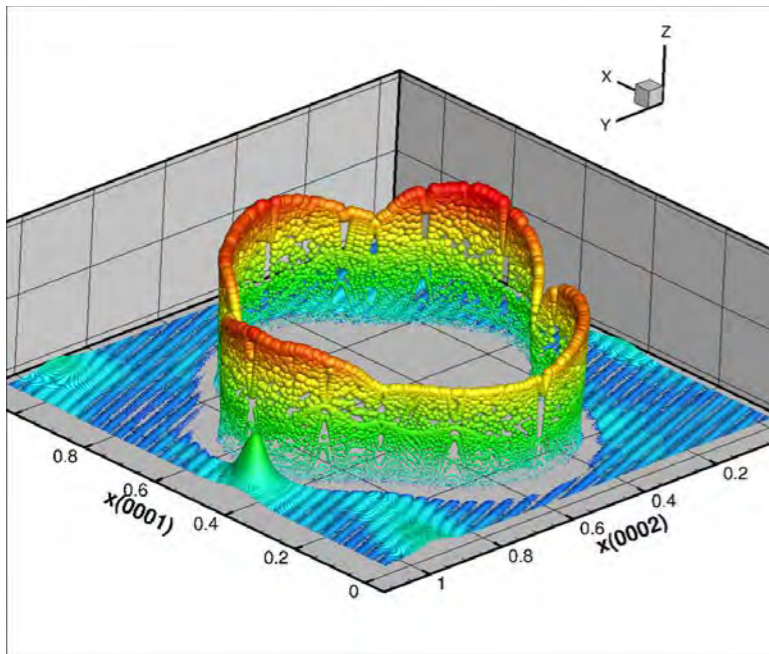
# Outcomes of Year 2

## Journal Articles

- GOAL: To predict “strong” and “weak” ignition phenomena in turbulent reacting flows with thermal inhomogeneities.

### Case A

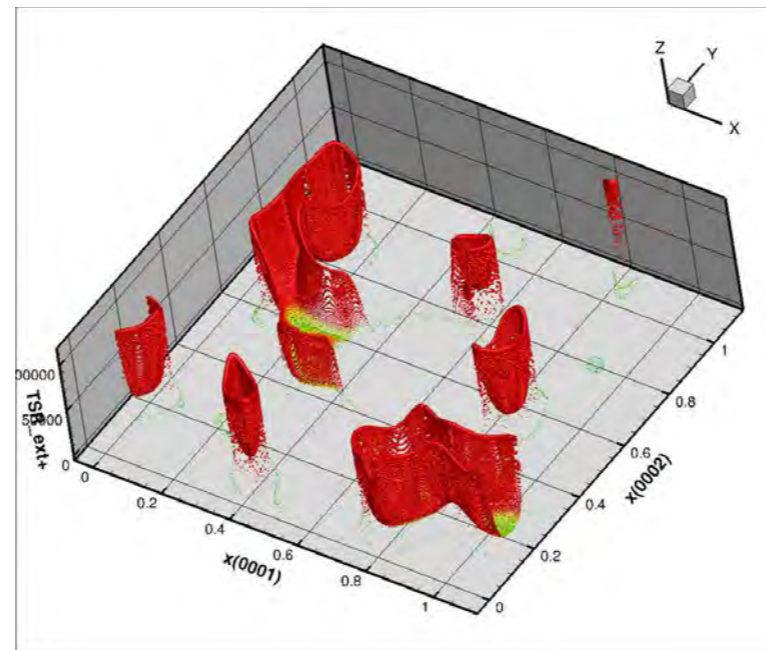
(Weak ignition)



Reaction front propagates by virtue of diffusion & convection: deflagration mode

### Case B

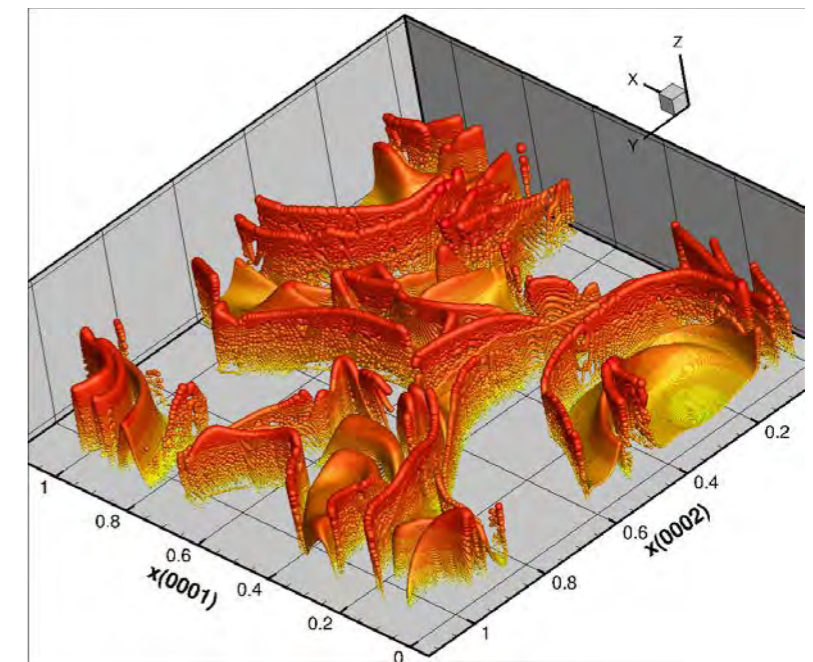
(Reaction-controlled strong ignition)



Reaction pockets occur by virtue of chemical reactions: spontaneous ignition mode

### Case C

(Mixing-controlled strong ignition)

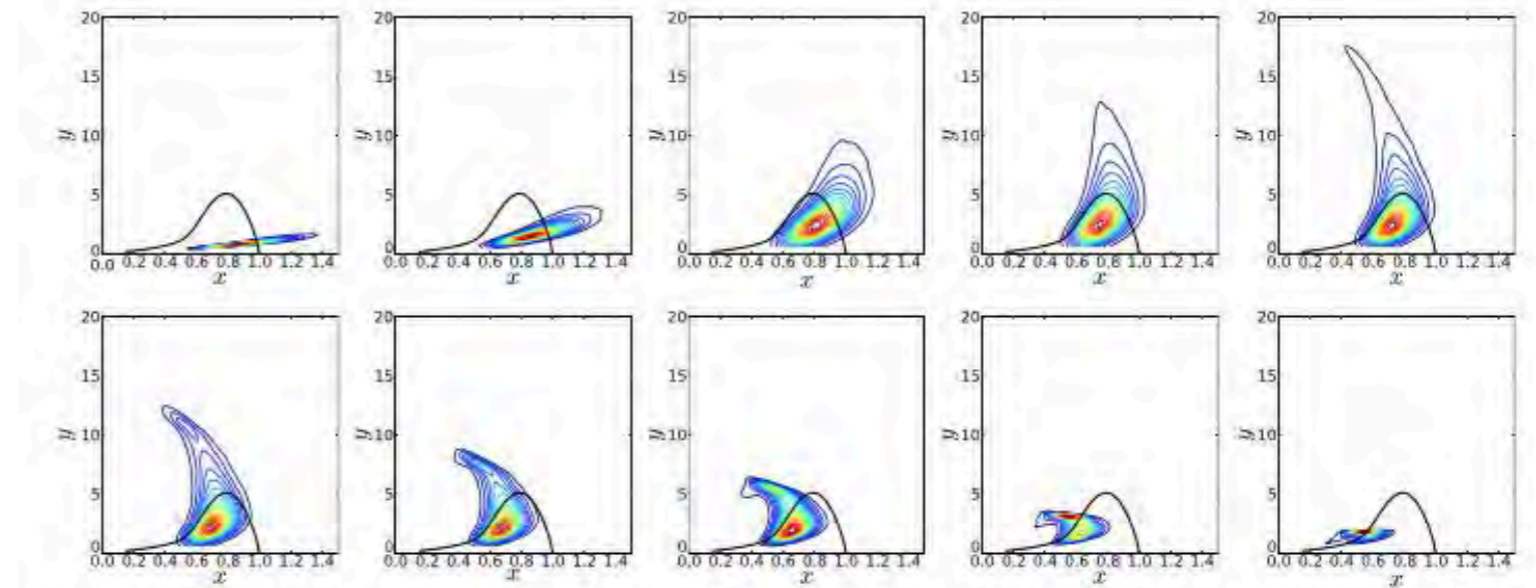
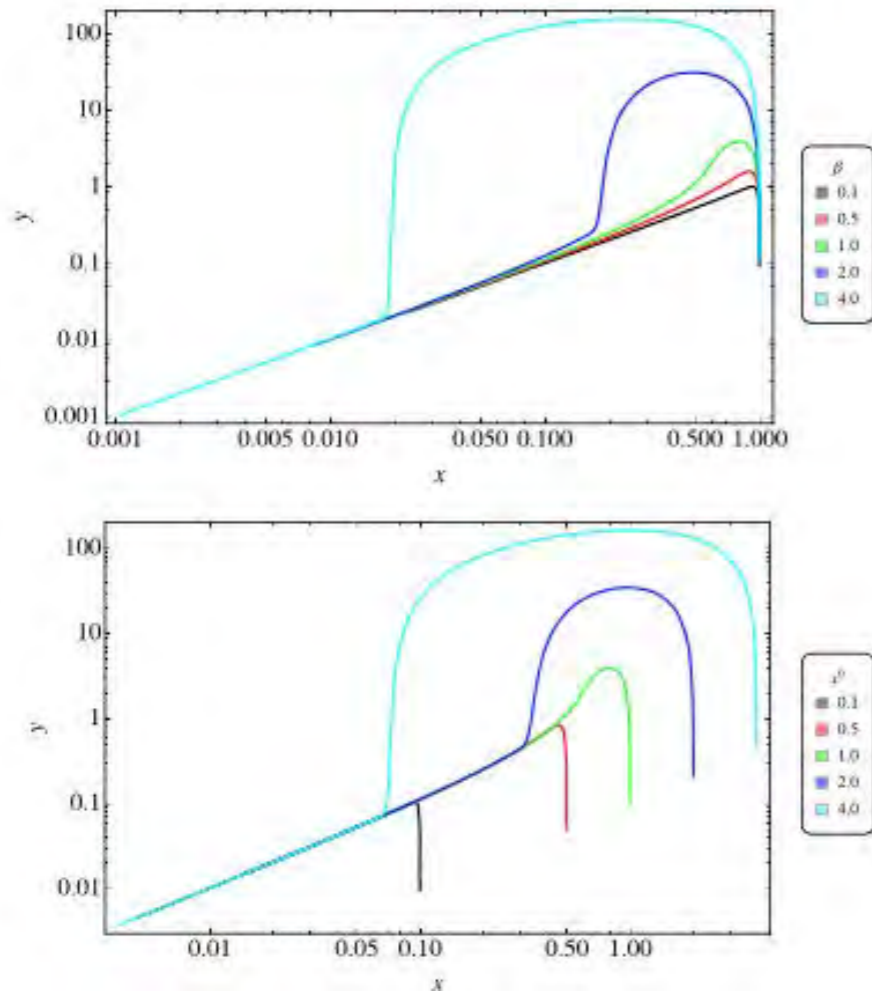


Reaction zone occur after mixing is completed: mixing-controlled strong ignition mode



# Uncertain Reactive ODE Systems

- H.N.Najm, M.Valorani, "Enforcing positivity in intrusive PC-UQ methods for reactive ODE systems", Journal of Computational Physics 270 (2014) 544–569



**Fig. 3.** Plot of the time evolution of  $p(x, y)$ . The frame time sequence is for  $t = 0.01, 0.02, 0.04, 0.06, 0.07, 0.08, 0.09, 0.10, 0.12, 0.15$ , ordered left to right and top to bottom. The frames are all on the same scale, with the horizontal axis showing  $x \in [0, 1.5]$ , and the vertical axis showing  $y \in [0, 20]$ . Using MC sampling, for the Case 1, superposed on the mean  $(x, y)$  trajectory.



# Chemical Model Reduction under Uncertainty

- R.Malpica Galassi, M.Valorani, H.N.Najm, C.Safta, M.Khalil, P.P.Ciottoli, "Chemical Model Reduction under Uncertainty", paper submitted to 36th Symposium Combustion 2016.

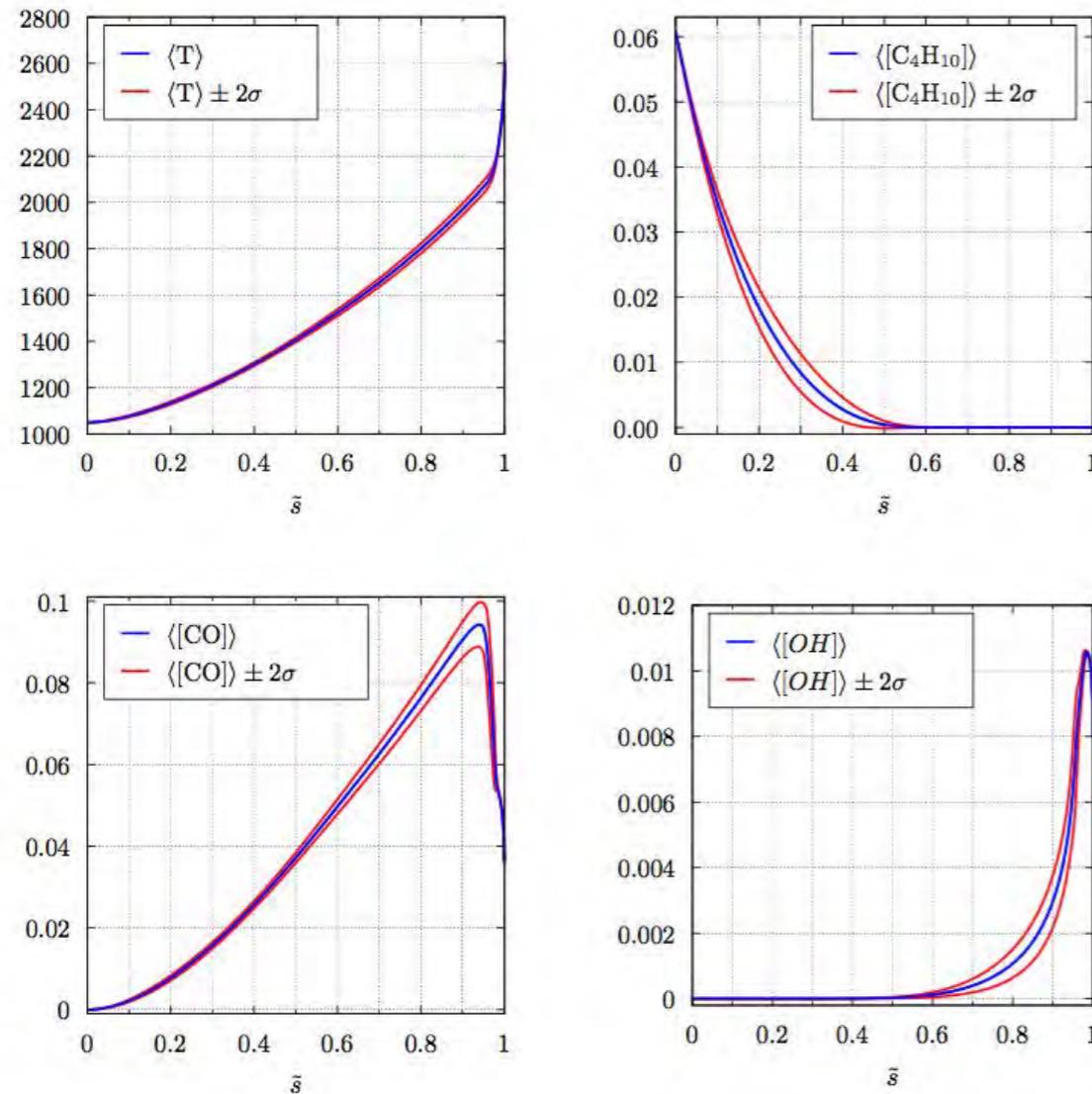


Figure 1: Average temperature,  $C_4H_{10}$ , CO and OH trajectories, with 2 standard deviation bounds

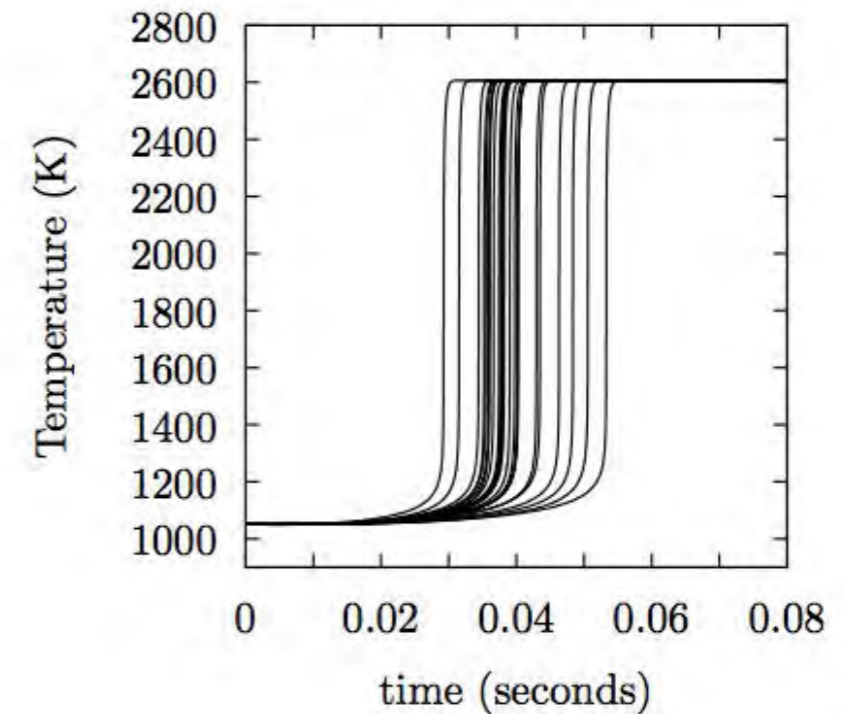


Figure 2: Temperature evolution in time for a number of samples



# Chemical Model Reduction under Uncertainty

- R.Malpica Galassi, M.Valorani, H.N.Najm, C.Safta, M.Khalil, P.P.Ciottoli, “Chemical Model Reduction under Uncertainty”, paper submitted to 36th Symposium Combustion 2016.

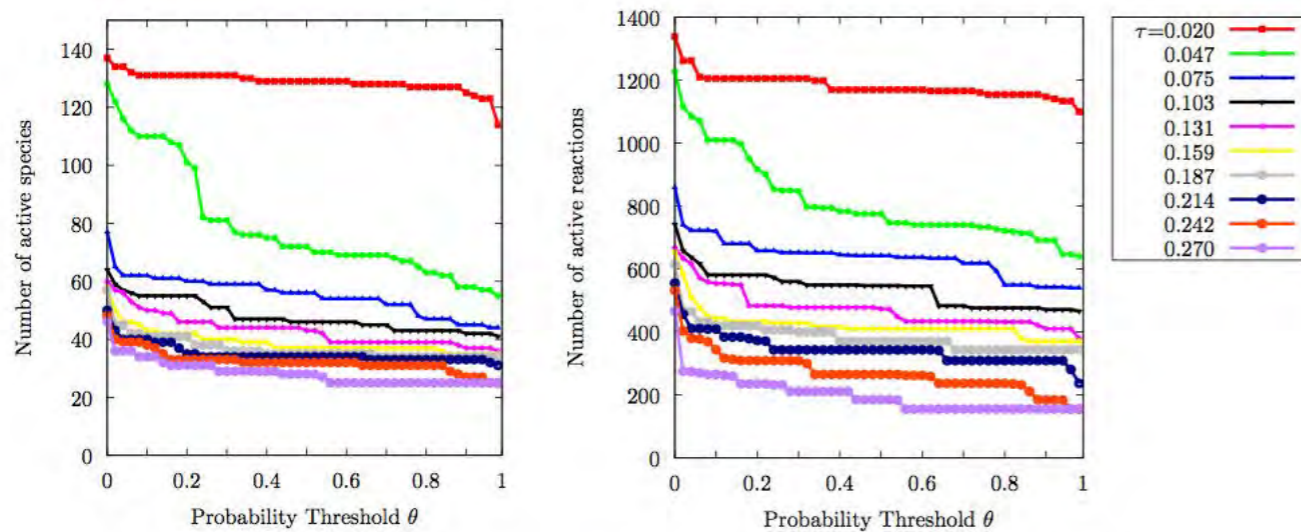
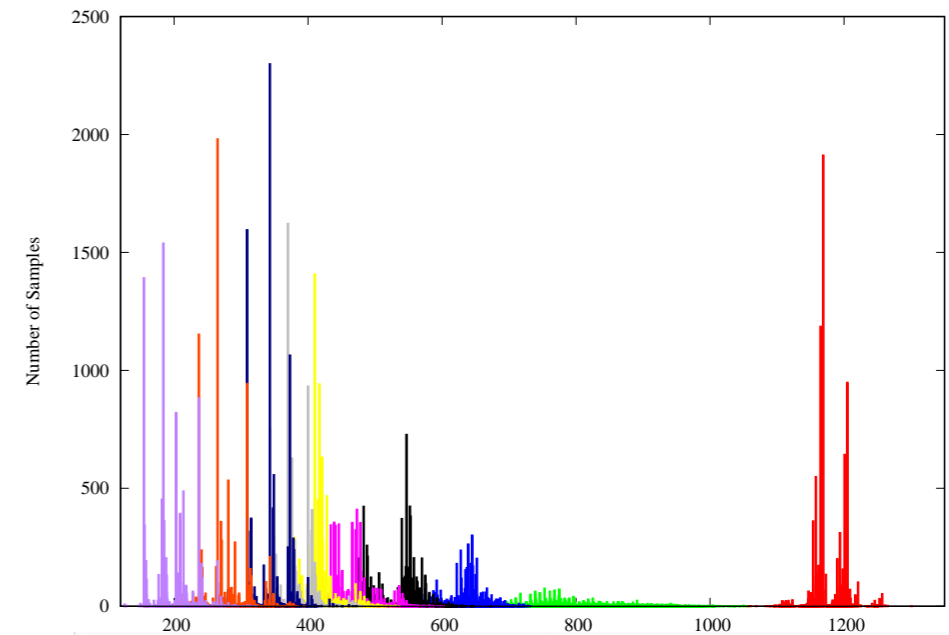
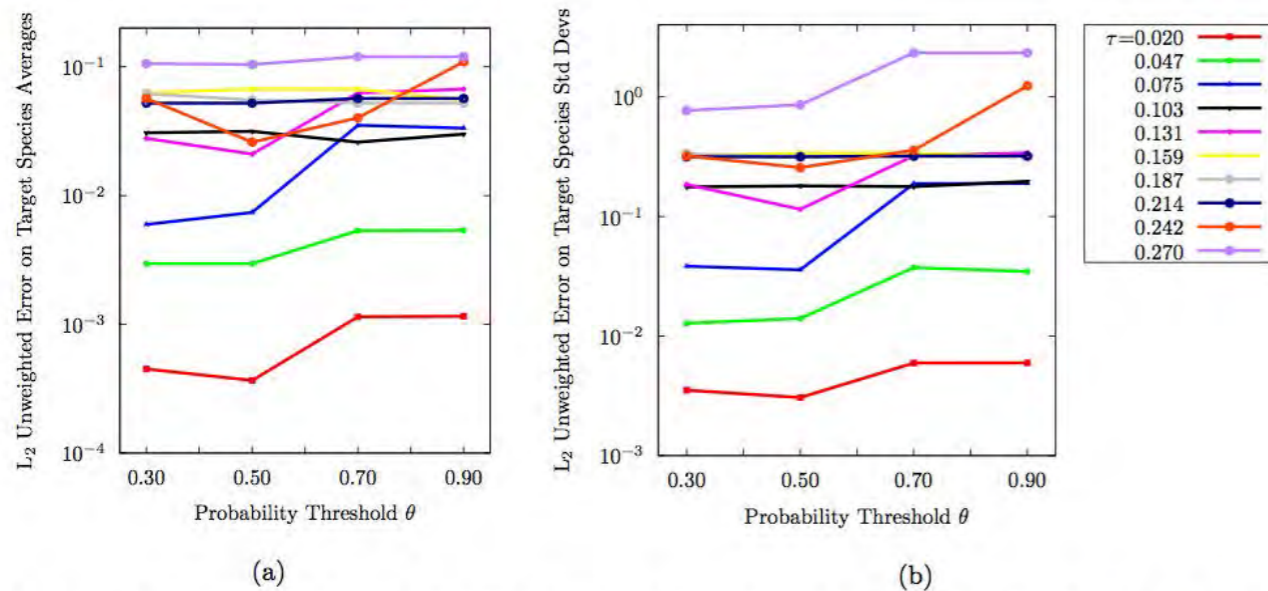


Figure 4: Number of retained/active species and reactions with increasing  $\tau$  and  $\theta$



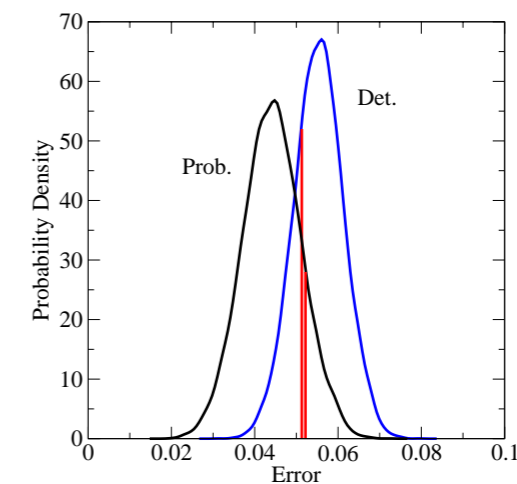
PDF of mechanism size at different threshold of importance



(a)

(b)

Figure 6: Average over the target species of the  $L_2$  un-weighted relative errors of the mean (left) and standard deviation (right) transient evolution against  $\theta$  thresholds for given  $\tau$  tolerances



Performance of deterministic vs probabilistic reduction



# Solvers for Systems of Stiff ODEs

- G-Scheme with re-use of CSP basis

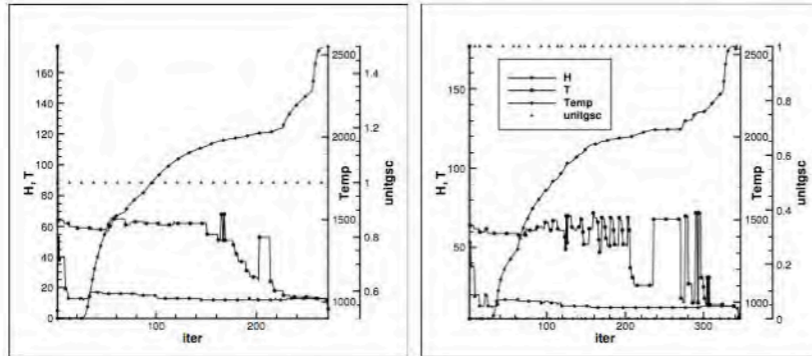


Figure 17: n-Butane. Without (left) and with (right) reuse of Jacobian and Basis. Evolution of head (square symbols) and tail (circle) dimensions; temperature evolution (diamond) and update(1)/reuse(0) (gradient)

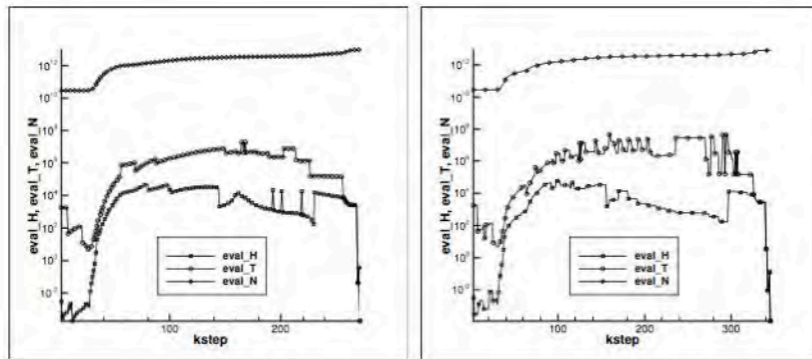


Figure 18: n-Butane. Without (left) and with (right) reuse of Jacobian and Basis. Evolution of head (square symbols), tail (circle), and fastest (diamond) time scales (reciprocal).

CPU Time: number of state variables  $N = N_s + 1$  (Temperature);  $rtol = 10^{-4}, 10^{-3}, 10^{-2}$

N vars	CVODE N Iters	GSC N Iters	CPU CVODE (s)	CPU GSC (s)	CPU GSC/CVODE
33	1161	233	0.05013	0.063311	1.26294
54	1310	251	0.095707	0.152625	1.59471
119	1675	298	0.290572	0.705717	2.42872
177	1615	281	0.489075	1.96473	4.01724
562	1434	308	4.65785	47.7425	10.2499

N vars	CVODE N Iters	GSC N Iters	CPU CVODE (s)	CPU GSC (s)	CPU GSC/CVODE
33	1161	200	0.05013	0.055073	1.0986
54	1310	200	0.095707	0.13264	1.3859
119	1675	217	0.290572	0.571197	1.96577
177	1615	203	0.489075	1.60425	3.28017
562	1434	227	4.65785	36.8613	7.9138

N vars	CVODE N Iters	GSC N Iters	CPU CVODE (s)	CPU GSC (s)	CPU GSC/CVODE
33	1161	592	0.05013	0.123645	2.46649
54	1310	497	0.095707	0.241588	2.52425
119	1675	323	0.290572	0.680094	2.34054
177	1615	376	0.489075	2.00814	4.10599
562	1434	327	4.65785	39.2042	8.4168

N vars	CVODE N Iters	GSC N Iters	CPU CVODE (s)	CPU GSC (s)	CPU GSC/CVODE
33	1161	619	0.05013	0.133874	2.67054
54	1310	563	0.095707	0.267126	2.79108
119	1675	656	0.290572	0.980779	3.37534
177	1615	598	0.489075	2.54091	5.19534
562	1434	411	4.65785	41.4817	8.90577





# Uniroma1 & NTUA Contribution

## How CSPTk runs

The screenshot shows a web browser window at localhost/csp/index.php displaying the 'List of Workflows' page. The page includes instructions for selecting a local folder and a table listing workflows. A file explorer window is open on the right, showing the contents of the 'pcost' folder.

**List of Workflows**

- 1. Select the local folder in which Workflows are stored
- 2. Select existing Workflow or create a new one

Select the local folder in which Workflows are stored:  
/Users/valorani/cspTk\_run/

**List of Workflows**

Add new Workflow

Name	Last Modification
✗ pcost	03-02-2017 13:09:51

File Explorer (pcost):  
gr13.0  
info.json

Footer: List of Workflows | CSP Documentation Support



# Uniroma1 & NTUA Contribution

## What comes out from CSPTk

The screenshot shows a web browser window with the URL `localhost/csp/workflow.php`. The page content includes:

- A list of steps: 1. Select mechanism, 2. Select simulation workflow and applicable actions, 3. Edit input values and run simulation, 4. Display simulation results.
- Form fields for configuration:
  - Select a mechanism:** `gri3.0`
  - Select a workflow:** `Complete ODE`
  - Model type:** `Constant volume`
  - Solver:** `cvode`
- Checkboxes for actions:
  - Create Data Base
  - Execute Analysis and Simplification
  - Execute Check
  - Compute Errors
  - Create Error Plots
- Edit input data:**
  - ▶ Initial conditions
  - ▶ Initial conditions for re-run
  - ▶ Simplify options
  - ▶ Solver options
  - ▶ Periodic table
- A `Run simulation` button.
- Results:**
  - Log files:**
    - ▶ Create Data Base
    - ▶ Execute Analysis and Simplification
    - ▶ Execute Check
    - ▶ Compute Errors
    - ▶ Create Error Plots
  - Plots:**
    - ▶ Create Data Base
    - ▶ Execute Analysis and Simplification
    - ▶ Execute Check
    - ▶ Create Error Plots

At the bottom, there is a breadcrumb `List of Workflows > Workflow` and links for `CSP Documentation` and `Support`.